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Taking Guide  
Episode 601  
Answers

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2004, GPB 6.27

Descriptive

Abbreviations ( S )

Solid ( l ) Liquid ( g

) Gas ( aq ) aqueous

Determining State  
at Room

Temperature • All  
metals are solids  
except for mercury,  
which is a liquid.

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Taking Guide

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chemical bond.  
ionic bond. cation.  
anion. forms when  
2 or more atoms  
rearrange valence  
electrons to incr....  
forms when  
valence electrons  
are transferred from  
one atom to a....

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atom loses  
electrons to  
become positively  
charged. atom  
gains electrons to  
become negatively  
charged.

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Quantum Numbers

$n, l, m, s$  Used to describe an

electron in an atom

$n$  Principle

quantum number

Represents main

energy level of

electron maximum

# of electrons in an

Energy level =

$2n^2$  Example:

What is the

maximum number

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of electrons that  
can be in the 5th  
main energy level?

l The 2<sup>nd</sup> quantum  
number •

Describes the  
orbital shape within  
an energy level •

number of orbital  
shapes possible in  
energy level =  $n$

Orbital Shapes

designated s, p ...



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f - orientation of  
orbital in space s ...

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and studies,  
either— these skills  
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## How Important Is Note-Taking?

According to  
research , human  
beings “tend to  
lose almost 40% of  
new information  
within the first 24  
hours” after first

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Answers  
give away

electrons, ions are  
positively charged,  
ions. nonmetals.

take electrons, ions  
are negatively  
charged, anions.

writing ionic

formulas. 1. write  
the symbols for the  
element. -the

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cation is written first. -The anion is written second. 2. Determine the charge on each ion.

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International

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## System (SI)

Measuring system  
used in science

AKA metric system

Length - distance  
between two points

Base unit- meter

(M) Instrument-

meter stick or

metric ruler

Mass

-amount of matter  
in an object. Base

unit- gram (G)

Instrument-

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## balance Volume

-amount of space  
an object occupies

Base unit-Liter (L)

Instrument- metric  
ruler or graduated  
cylinder Time -

interval between  
two occurrences

Base unit- seconds  
(S) Instrument ...

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